$$\begin{array}{c} O \xrightarrow{1. \text{ K}^+\text{N}_3^-} \\ O \xrightarrow{2. \text{ H}_2\text{O}} & O \\ O H \end{array} + \begin{array}{c} O H \\ \hline \\ O H \\ \hline \end{array} \begin{array}{c} O H \\ \hline \\ O H \\ \hline \end{array} \begin{array}{c} O H \\ \hline \\ O H \\ \hline \end{array} \begin{array}{c} O H \\ \hline \\ O H \\ \hline \end{array} \begin{array}{c} O H \\ \hline \\ O H \\ \hline \end{array} \begin{array}{c} O H \\ \hline \\ O H \\ \hline \end{array} \begin{array}{c} O H \\ \hline \\ O H \\ \hline \end{array} \begin{array}{c} O H \\ \hline \\ O H \\ \hline \end{array} \begin{array}{c} O H \\ \hline \\ O H \\ \hline \end{array} \begin{array}{c} O H \\ \hline \\ O H \\ \hline \end{array} \begin{array}{c} O H \\ \hline \\ O H \\ \hline \end{array} \begin{array}{c} O H \\ \hline \\ O H \\ \hline \end{array} \begin{array}{c} O H \\ \hline \\ O H \\ \hline \end{array} \begin{array}{c} O H \\ \hline \\ O H \\ \hline \end{array} \begin{array}{c} O H \\ \hline \\ O H \\ \hline \end{array} \begin{array}{c} O H \\ \hline \\ O H \\ \hline \end{array} \begin{array}{c} O H \\ \hline \\ O H \\ \hline \end{array} \begin{array}{c} O H \\ \hline \\ O H \\ \hline \end{array} \begin{array}{c} O H \\ \hline \\ O H \\ \hline \end{array} \begin{array}{c} O H \\ \hline \\ O H \\ \hline \end{array} \begin{array}{c} O H \\ \hline \\ O H \\ \hline \end{array} \begin{array}{c} O H \\ \hline \\ O H \\ \hline \end{array} \begin{array}{c} O H \\ \hline \\ O H \\ \hline \end{array} \begin{array}{c} O H \\ \hline \\ O H \\ \hline \end{array} \begin{array}{c} O H \\ \hline \\ O H \\ \hline \end{array} \begin{array}{c} O H \\ \hline \\ O H \\ \hline \end{array} \begin{array}{c} O H \\ \hline \\ O H \\ \hline \end{array} \begin{array}{c} O H \\ \hline \\ O H \\ \hline \end{array} \begin{array}{c} O H \\ \hline \\ O H \\ \hline \end{array} \begin{array}{c} O H \\ \hline \\ O H \\ \hline \end{array} \begin{array}{c} O H \\ \hline \\ O H \\ \hline \end{array} \begin{array}{c} O H \\ \hline \\ O H \\ \hline \end{array} \begin{array}{c} O H \\ \hline \\ O H \\ \hline \end{array} \begin{array}{c} O H \\ \hline \\ O H \\ \hline \end{array} \begin{array}{c} O H \\ \hline \\ O H \\ \hline \end{array} \begin{array}{c} O H \\ \hline \\ \end{array} \begin{array}{c} O H \\ \hline \\ O H \\ \hline \end{array} \begin{array}{c} O H \\ \hline \\ O H \\ \hline \end{array} \begin{array}{c} O H \\ \hline \end{array} \begin{array}{$$

## **Gabriel Synthesis**

$$R-X + K^{+} \cdot N$$

$$R-NH_{2}$$

Potassium phthalimide is a "NH<sub>2</sub>-synthon which allows the preparation of primary amines by reaction with alkyl halides. After alkylation, the phthalimid is not nucleophile and does not react anymore. Product is cleaved by reaction with base or hydrazine, which leads to a stable cyclic product.

## Mechanism of the Gabriel Synthesis

Note: Phthalimide is acidic!

Cleavage:

**14. Sandmeyer Reaction (Section 23.8E)** Treatment of an arenediazonium salt with CuCl, CuBr, or CuCN results in replacement of the diazonium group by —Cl, —Br, or —CN, respectively.

$$CH_3 \longrightarrow NH_2 \longrightarrow NaNO_2, H_3O^+ \longrightarrow N_2^+ \longrightarrow N_2^+ \longrightarrow CH_3$$

$$N_2 \longrightarrow N_2 \longrightarrow N_2^+ \longrightarrow N_$$