# Chemistry 2541, Fall 2017 Midterm Exam 3 

(100 points)

## Important notes:

- Please use the provided Scantron form for your answers; you can keep the sheet with the questions and can use it as scratch paper
- Do not forget to write your name on the Scantron form
- You will not receive credit for unmarked answers or for more than one mark on answer line
- Your scores will be posted on eGradebook; graded Scantron forms will not be returned to students.

Questions 1-28 (84 pts): Please mark the appropriate box on the front of the Scantron form (3 pts each).

1. What is the IUPAC name for the compound shown in the box?

A) (E)-2-methyl-5-pentyn-1-ene
B) 2-methyl-1-penten-4-yne
C) 2-methyl-1-hexen-4-yne
D) 4-methyl-1-hexyn-4-ene
2. What is the IUPAC name for the compound shown in the box?

A) 3-pentyn-2-ol
B) 2-pentyn-4-ol
C) 2-methyl-3-butynol
D) 1-methyl-3-butyn-2-ol
3. What is the IUPAC name of the major product for the reaction shown in the box?

$\xrightarrow{\text { 1) } \mathrm{Br}_{2}}$ ?
2) $\mathrm{NaNH}_{2}$ (excess)
3) $\mathrm{H}_{2} \mathrm{O}$
A) 4,4-dimethyl-2-pentyne
B) 3,3-dimethyl-1-butyne
C) 4,4-dimethyl-1-pentyne
D) 3,3-dimethyl-1-pentyne
4. A terminal alkyne, 1-butyne is NOT deprotonated by the ethoxide ion. What does this indicate?
A) Alkynyl anion is a weaker base than the hydroxide ion
B) 1-Butyne is stronger base than ethanol
C) Ethanol is a stronger acid than 1-butyne
D) 1-Butyne is the conjugate base of ethanol
5. What is the IUPAC name of the major product for the reaction shown in the box?

A) 2-methyl-1-hepten-5-yne
B) (E)-3-methyl-2-hepten-5-yne
C) 2-methyl-2-heptyne
D) (Z)-3-methyl-2-hepten-5-yne
6. What is a main product of the reaction shown in the box?
?
A)

B)

C)

D)

7. What is the IUPAC name of the major product for the reaction sequence shown in the box?
$\mathrm{CH}_{3} \mathrm{C} \equiv \mathrm{CH} \xrightarrow{\mathrm{NaNH}_{2}} \xrightarrow{\mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{Br}} \xrightarrow[\mathrm{CH}_{2} \mathrm{Cl}_{2}]{1 \mathrm{~mol} \mathrm{Br}_{2}} \boldsymbol{?}$
A) (E)-2,3-dibromo-2-pentene
B) 2,3-dibromopentane
C) (Z)-2,3-dibromo-2-pentene
D) 3,3-dibromopentane
8. What is the main product of the reaction shown in the box?

A)

B)

C)

D)

9. What is the main product of the reaction sequence shown in the box?

A)
$\square=$
B)

C)

D)

10. What is the main product of the reaction shown in the box?
$\mathrm{CH}_{3} \mathrm{C} \equiv \mathrm{CH}+\mathrm{H}_{2} \mathrm{O} \xrightarrow[\mathrm{HgSO}_{4} \text { (cat) }]{\mathrm{H}_{2} \mathrm{SO}_{4} \text { (cat) }}$ ?
A) $\mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{OH}$
B) $\mathrm{CH}_{3} \mathrm{CH}(\mathrm{OH}) \mathrm{CH}_{3}$
C) $\mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{CHO}$
D) $\mathrm{CH}_{3} \mathrm{COCH}_{3}$
11. What is the main product of the reaction shown in the box?

A)

B)

C)

D)

12. Which sequence of reagents can be used for the reaction shown in the box?

A) (1) $\mathrm{CH}_{3} \mathrm{Br}$, (2) $\mathrm{BH}_{3}$
B) (1) $\mathrm{CH}_{2} \mathrm{CH}_{3} \mathrm{Br}$, (2) $\mathrm{BH}_{3}$
C) (1) $\mathrm{CH}_{3} \mathrm{Br}$, (2) $\mathrm{H}_{2} \mathrm{SO}_{4}$
D) (1) $\mathrm{CH}_{2} \mathrm{CH}_{3} \mathrm{Br},(\mathbf{2}) \mathrm{HgSO}_{4}$
13. Which sequence of reagents can be used for the reaction shown in the box?

A) (1) NaH , (2) $\mathrm{C}_{2} \mathrm{H}_{5} \mathrm{Br}$, (3) $\mathrm{H}_{2} \mathrm{SO}_{4}$
B) (1) $\mathrm{NaNH}_{2}$, (2) $\mathrm{CH}_{3} \mathrm{Br}$, (3) $\mathrm{H}_{2} \mathrm{SO}_{4}$
C) (1) $\mathrm{Br}_{2}$ (1), (2) $\mathrm{NaNH}_{2}$, (3) $\mathrm{BH}_{3}$
D) (1) $\mathrm{NaNH}_{2}$, (2) $\mathrm{CH}_{3} \mathrm{Br}$, (3) $\mathrm{BH}_{3}$
14. Which sequence of reagents can be used for the reaction shown in the box?

15. Which one of the following compounds has high solubility in water?
A) $\mathrm{CHBr}_{3}$
B)

C)

D)

16. What is the main product of the reaction shown in the box?

$\xrightarrow{\mathrm{Na}}$ ?
A)

B)

C)
D)


17. What is the main product of the reaction shown in the box?

A)

B)

C)

D)

18. What is the IUPAC name of the major product for the reaction sequence shown in the box?

|  | 1. $\mathrm{OsO}_{4}$ | 1. NaH (excess) |
| :---: | :---: | :---: |
|  | 2. $\mathrm{NaHSO}_{3}, \mathrm{H}_{2} \mathrm{O}$ | 2. $\mathrm{CH}_{3} \mathrm{I}$ (excess) |

A) cis-1,2-dimethoxycyclohexane
B) cis-1,2-dimethylcyclohexane
C) trans-1,2-dimethoxycyclohexane
D) trans-1,2-dimethylcyclohexane
19. What is the IUPAC name of the major product for the reaction shown in the box?

A) (E)-1-bromo-1,3-dichloro-1-butene
B) (Z)-1-bromo-1,3-dichloro-1-butene
C) (E)-1-bromo-1,3-dichloro-2-butene
D) (Z)-1-bromo-1,3-dichloro-2-butene
20. What is the IUPAC name of the major product for the reaction shown in the box?

A) $(2 S, 3 S)$-2,3-dibromobutane
C) (R)-1,2-dibromobutane
B) $(2 R, 3 S)$-2,3-dibromobutane
D) (S)-1,2-dibromobutane
21. What is the main product of the reaction shown in the box?

A) $\mathrm{PhCH}_{2} \mathrm{CH}_{2} \mathrm{CN}$
B) $\mathrm{PhCH}_{2} \mathrm{CN}$
C)
D)

22. What is the IUPAC name of the major product for the reaction shown in the box?

A) 1-methylcyclohexene
B) 3-methylcyclohexene
C) 4-methylcyclohexene
D) methylenecyclohexene
23. What is the main product of the reaction shown in the box?
$\xrightarrow[2 . \mathrm{H}_{2} \mathrm{O}_{2}, \mathrm{NaOH}, \mathrm{H}_{2} \mathrm{O}]{\stackrel{\mathrm{BH}_{3}}{\text { ? }} \xrightarrow[\text { pyridine }]{\mathrm{TsCl}} \text { ? }}$
A)

B)

C)

D)


24 . What is the main product of the reaction shown in the box?
HO

$\xrightarrow[\text { 2) } \mathrm{NaSH} \text {, acetone }]{\text { 1) } \mathrm{TsCl} \text {, pyridine }}$ ?
A) $\mathrm{HO}_{1}$

B) HO
C)

D)

25. Which sequence of reagents can be used for the reaction shown in the box?

A) (1) $\mathrm{NaN}_{3}$, (2) IBX
B) (1) IBX, (2) $\mathrm{NaN}_{3}$
C) (1) TsCl, (2) $\mathrm{NaN}_{3}$
D) (1) $\mathrm{NaN}_{3}$, (2) TsCl
26. What is the main product of the reaction sequence shown in the box?

A)

B)

C)

D)

27. Which reagent can be used for the reaction shown in the box?



A) IBX
B) $\mathrm{H}_{2} \mathrm{CrO}_{4}$
C) PCC
C) NBS
28. Which sequence of reagents can be used for the reaction shown in the box?

A) (1) NaOH , (2) IBX
B) (1) $\mathrm{CH}_{3} \mathrm{ONa}$, (2) PCC
C) (1) NBS, (2) PCC
D) (1) NaOH , (2) $\mathrm{H}_{2} \mathrm{CrO}_{4}$

Question 29: Provide the reagents that give the indicated products in high yield. Please write your answers in boxes 66-69 on the back of the Scantron form (4 pts each).








