## Chemistry 2541, Fall 2017 Quiz 2

(30 points)

Important notes:

- Please use the provided Scantron form for your answers; you can keep the sheet with the questions and can use it as scratch paper
- Do not forget to write your name on the Scantron form
- You will not receive credit for unmarked answers or for more than one mark on answer line
- Your scores will be posted on eGradebook; graded Scantron forms will not be returned to students.

Questions 1-10: Please mark the appropriate box on the front of the Scantron form (3 pts each).

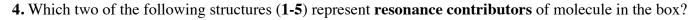
1. Which one of the following pairs of structures depicts **resonance contributors**?

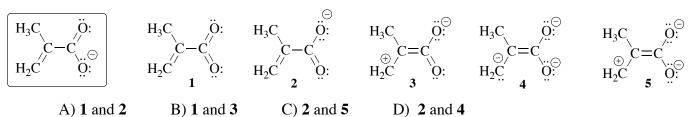
A) 
$$\vdots \overset{\bigcirc}{\circ} : \overset{\bigcirc}{\circ} : \overset{\circ}{\circ} - H$$
  $\vdots \overset{\circ}{\circ} - H$   $\vdots \overset{\circ$ 

2. Which one of the following structures represents major resonance contributor of molecule in the box?

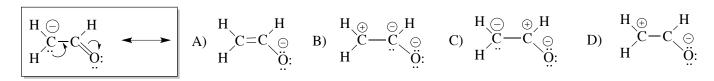
$$\begin{bmatrix} H_3C & O \\ \vdots & N-C & \ominus \\ H_2C & O \\ \end{bmatrix} \quad A) \quad \begin{matrix} H_3C & O \\ N-C & B \end{matrix} \quad \begin{matrix} \vdots \\ N-C & O \\ \end{matrix} \quad B) \quad \begin{matrix} M-C & O \\ N-C & C \end{matrix} \quad \begin{matrix} \vdots \\ N-C & O \\ \end{matrix} \end{matrix} \quad \begin{matrix} \vdots \\ N-C & O \\ \end{matrix} \end{matrix} \quad \begin{matrix} \vdots \\ N-C & O \\ \end{matrix} \end{matrix} \quad \begin{matrix} \vdots \\ N-C & O \\ \end{matrix} \end{matrix} \quad \begin{matrix} \vdots \\ N-C & O \\ \end{matrix} \end{matrix} \quad \begin{matrix} \vdots \\ N-C & O \\ \end{matrix} \end{matrix} \quad \begin{matrix} \vdots \\ N-C & O \\ \end{matrix} \end{matrix} \quad \begin{matrix} \vdots \\ N-C & O \\ \end{matrix} \end{matrix} \end{matrix} \quad \begin{matrix} \vdots \\ N-C & O \\ \end{matrix} \end{matrix} \begin{matrix} \begin{matrix} \vdots \\ N-C & O \\ \end{matrix} \end{matrix} \end{matrix} \begin{matrix} \begin{matrix} \vdots \\ N-C & O \\ \end{matrix}$$

3. Which one of the following structures represents major resonance contributor of molecule in the box?

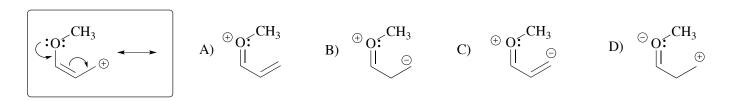




**5.** Which one of the following structures represents **resonance contributor** of molecule in the box in agreement with the shown curved **arrows**?



**6.** Which one of the following structures represents **resonance contributor** of molecule in the box in agreement with the shown curved **arrows**?



- **7.** Which **orbitals** overlap to form the carbon-carbon  $\sigma$  **bond** of acetic acid, CH<sub>3</sub>CO<sub>3</sub>H?
- A)  $\mathbf{sp}^3 + \mathbf{sp}^5$  B)  $\mathbf{sp}^2 + \mathbf{sp}^2$  C)  $\mathbf{sp}^3 + \mathbf{sp}^2$  D)  $\mathbf{sp} + \mathbf{sp}^3$
- **8.** Which one of the following compounds has *sp*-hybridized atoms in the structure?

A) 
$$CH_3C \equiv N$$
:

B)  $CH_3C = CH_2$ 
 $CH_3C = CH_3$ 
 $CH_3C = CH_3$ 

**9.** Which two the following compounds (molecules **1-5**) have **two**  $\pi$  bonds in their structures?

- 10. Which one of the following molecules has the shortest carbon-carbon bond?
- A)  $C_2H_4$  B)  $CH_3CH=CH_2$  C)  $C_2H_6$  D)  $C_2H_2$