# Chemistry 2541, Fall 2017 <br> Quiz 2 

(30 points)

## Important notes:

- Please use the provided Scantron form for your answers; you can keep the sheet with the questions and can use it as scratch paper
- Do not forget to write your name on the Scantron form
- You will not receive credit for unmarked answers or for more than one mark on answer line
- Your scores will be posted on eGradebook; graded Scantron forms will not be returned to students.

Questions 1-10: Please mark the appropriate box on the front of the Scantron form (3 pts each).

1. Which one of the following pairs of structures depicts resonance contributors?
A)

and

B)


C)


and

D)

 and
and



2. Which one of the following structures represents major resonance contributor of molecule in the box?


B)

C)

D)

3. Which one of the following structures represents major resonance contributor of molecule in the box?

A)

B)

C)

D)

4. Which two of the following structures (1-5) represent resonance contributors of molecule in the box?


B) 1 and 3




D) 2 and 4
5. Which one of the following structures represents resonance contributor of molecule in the box in agreement with the shown curved arrows?

A)

B)

C)

D)

6. Which one of the following structures represents resonance contributor of molecule in the box in agreement with the shown curved arrows?

A)

B)

C)

D)

7. Which orbitals overlap to form the carbon-carbon $\sigma$ bond of acetic acid, $\mathrm{CH}_{3} \mathrm{CO}_{2} \mathrm{H}$ ?
A) $\mathbf{s p}^{3}+\mathbf{s p}$
B) $\mathbf{s p}+\mathbf{s p}$
C) $\mathbf{s p}^{3}+\mathbf{s} \mathbf{p}^{2}$
D) $\mathbf{s p}+\mathbf{s p}$
8. Which one of the following compounds has $\boldsymbol{s p}$-hybridized atoms in the structure?
A) $\mathrm{CH}_{3} \mathrm{C} \equiv \mathrm{N}$ :
B)

C)

D) $\mathrm{C}_{2} \mathrm{H}_{4}$
9. Which two the following compounds (molecules $\mathbf{1 - 5}$ ) have two $\pi$ bonds in their structures?


$\mathrm{C}_{2} \mathrm{H}_{2} \quad \mathrm{C}_{2} \mathrm{H}_{4}$
$4 \quad 5$
A) $\mathbf{1}$ and $\mathbf{2}$
B) 2 and 4
C) $\mathbf{3}$ and $\mathbf{4}$
D) $\mathbf{3}$ and 5
10. Which one of the following molecules has the shortest carbon-carbon bond?
A) $\mathrm{C}_{2} \mathrm{H}_{4}$
B) $\mathrm{CH}_{3} \mathrm{CH}=\mathrm{CH}_{2}$
C) $\mathrm{C}_{2} \mathrm{H}_{6}$
D) $\mathrm{C}_{2} \mathrm{H}_{2}$
