## Chemistry 2541

## Fall Semester 2010

## Quiz 2

(25 points)
September 22, 2010

Printed Name (Last, First)

Good Luck!

1. (4) Answer the following four questions about the molecule shown in the box below (write numbers after each question; 1 pt each answer):


- Number of non-bonding electrons present in this molecule:
- Number of $s p^{2}$-hybridized atoms in this molecule:
- Number of $\pi$ bonds in this molecule:
- Number of $\sigma$ bonds in this molecule:

2. (3) Which orbitals overlap to form the carbon-oxygen $\sigma$ bond of acetaldehyde, $\mathbf{C H}_{3} \mathbf{C H O}$ ?

$$
\text { (circle correct answer): } \quad \mathbf{s p}^{3}+\mathbf{s p}^{3} \quad \mathbf{s p}^{2}+\mathbf{s p}^{2} \quad \mathbf{s p}^{3}+\mathbf{s p}^{2} \quad \mathbf{s p}+\mathbf{s p ^ { 2 }} \quad \mathbf{s p}+\mathbf{s p}^{3}
$$

3. (6) In the provided boxes, finish drawing of the most important resonance contributing structures for each of the following species by placing missing bonds or formal charges at appropriate position. (3 pts each structure; no partial credit)

4. (6) Similarly to the previous problem, finish drawings of the resonance contributing structures which are in agreement with the shown curved arrows. (3 pts each structure; no partial credit)


5. (3) Which one of the following molecules has the longest $\mathbf{C}-\mathbf{H}$ bond?

6. (3) Circle the correct line-angle structure for the molecule shown in the box:






Overall Score: $\qquad$

