## Chemistry 2542

# Fall Semester 2012; Midterm 1 Exam 

October 10, Wednesday, 11:00 to 11:50 am

This exam has 6 problems ( 100 pts ) on 8 pages. Make sure your copy is complete and correct.

Printed Name (Last, First) $\qquad$

Your graded exams will be available Friday, October 12, before class.

## Chemistry 2542 <br> Fall 2012; Midterm 1 Exam

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Printed Name (Last, First)

## Scores:

Problem 1: $\qquad$

Problem 2: $\qquad$

Problem 3: $\qquad$

Problem 4: $\qquad$

Problem 5: $\qquad$

Problem 6: $\qquad$

Total: $\qquad$

1. ( $\mathbf{1 5} \mathbf{~ p t s}$ ) Answer the questions on mechanisms of the following reactions.
(a) (5 pts) Which one of the following four schemes (A-D) gives the best representation of a step in the mechanism of the reaction in the box (circle the correct answer):

(b) (5 pts) Which one of the following four schemes (A-D) gives the best representation of a step in the mechanism of the reaction in the box (circle the correct answer):


(c) ( 5 pts) Draw 4 curved arrows and show 1 charge missing in the following mechanism (1 pt each arrow or charge):

2. (16 pts) Answer the following questions.
(a) (4 pts) Circle the structure of the intermediate for the reaction in the box:


(b) (4 pts) Which one of the following compounds is the enol form of 3-oxopropanal?





(c) ( $\mathbf{8} \mathbf{p t s}$ ) Circle the correct IUPAC name of the compounds in the boxes ( 2 pts each):
$\mathrm{H}_{3} \mathrm{CO}_{\mathrm{m}}$. OH
cis-3-methoxycyclopentanol cis-1-hydroxy-3-methoxycyclopentane trans-3-methoxycyclopentanol trans-1-hydroxy-3-methoxycyclopentane

(E)-4-hydroxy-2-butenal
(Z)-4-hydroxy-2-butenal
(E)-1-hydroxy-2-buten-4-al
(Z)-1-hydroxy-2-buten-4-al
(E)-4-oxo-2-butenol (Z)-4-oxo-2-butenol


1-phenylcyclopentane epoxide 2-phenylcyclopenteneoxirane
1-phenyl-1,2-epoxycyclopentane 1-phenylcyclopentaneoxide
phenylcarbaldehyde benzaldehyde benzophenone benzenal
3. ( $\mathbf{2 6} \mathbf{~ p t s}$ ) Answer the following questions:
(a) Circle the name of a major product in each of the following reactions ( $\mathbf{5} \mathbf{~ p t s ~ e a c h ) : ~}$

| $\mathrm{H}_{3} \mathrm{C}_{m}$  | $\underset{\text { 2. } \mathrm{CH}_{3} \mathrm{I}}{\text { 1. } \mathrm{NaH}} \text { ? }$ | trans-1-methoxy-3-methylcyclopentane 3-methoxycyclopentene trans-3-methoxycyclopentanol cis-3-methoxycyclopentanol <br> 3-methylcyclopentene cis-1-methoxy-3-methylcyclopentane |
| :---: | :---: | :---: |




(b) ( 6 pts) Finish drawing the main product for each the following reactions by placing appropriate substituents in the boxes; 1 pt each.

4. (21 pts) (a) Circle the structure of a major product in each of the following reactions (4 pts each):











$\mathrm{HO}_{2} \mathrm{CCH}_{2} \mathrm{CO}_{2} \mathrm{H}$
(b) ( 5 pts ) Which pair of reactants is required to synthesize the compound in the box (circle the answer):

(A) $\underbrace{}_{\mathrm{O}}+\mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{Li}$
(C)

(B)

(D)

5. ( $\mathbf{1 0} \mathbf{~ p t s}$ ) Answer the following spectroscopy questions:
(a) (4 pts) Circle the structure of the compound which has a broad, strong IR absorption at 3200-3500 $\mathrm{cm}^{-1}$ and the following ${ }^{1} \mathrm{H}$ NMR spectrum:

(b) ( $\mathbf{2} \mathbf{~ p t s}$ ) How many signals would you expect to observe in the ${ }^{13} \mathrm{C}$ NMR spectra of the molecule of 1,3-cyclohexanedione (put the number in the box):

(c) ( $\mathbf{2} \mathbf{~ p t s}$ ) Which of the following compounds will have the characteristic IR peak at about $2150 \mathrm{~cm}^{-1}$ and the molecular peak $\mathrm{M}^{+}=44$ in the mass spectrum (atomic weight of C is $12, \mathrm{~N} 14, \mathrm{~F} 19, \mathrm{H} 1$ )?

$$
\mathrm{CH}_{3}-\mathrm{C} \equiv \mathrm{~N} \quad \mathrm{CH}_{3}-\mathrm{C} \equiv \mathrm{CH} \quad \mathrm{~F}-\mathrm{C} \equiv \mathrm{CH} \quad D \quad \mathrm{NH}_{2}-\mathrm{C} \equiv \mathrm{CH} \quad \mathrm{O}=
$$

(d) ( $\mathbf{2} \mathbf{~ p t s}$ ) Which of the following compounds will have the most deshielded carbon atom? (Circle one structure)

6. (12 pts) Place in each box the molecule of a reagent that is required to perform each of the following reactions (2 pts each box):






