

A. Notebook. General organization, completeness, documentation, clarity.	Comments
1. Table of Contents up-to-date	
2. Title, dates, signature	
3. Summary of method, reactions	
4. Stepwise working procedure	
5. Data entered in ink, documented. Original data reorganized as needed to produce original, legible record.	
B. Notebook. Calculations and Data Treatment	
1. Data complete	
2. Equations and calculations	
3. Error analysis	
C. Lab Practice	
1. Wears safety goggles at all times	
2. Practices good chemical hygiene (keeps clean, organized work space at workstation and elsewhere in lab)	
3. Arrives on time for beginning of lab	
4. Employs recommended lab techniques (eg, weighing by difference, data entered directly in notebook)	
5. Comes to lab prepared (eg, procedure outlined in notebook)	
General Comments/Recommendations	
Notebook and Lab Practice Grade (10)	
D. Accuracy of Reported Result (100)	

Detail for Part B

1. Data Complete
 - a. Complete minimum of 3 trials total, including one or more potentiometric titrations.
 - b. Plot pH vs volume of std HCl
2. Data and calculations
 - a. Values of K_{a1} and K_{a2} for carbonic acid from titration curve.
 - b. Equation for %Na₂CO₃ in unknown with data entered, mean reported.
3. Error Analysis
 - a. Discuss effects of carbonate equilibria on location of equivalence points.
 - b. Standard deviation and 90% confidence limits on average %Na₂CO₃.