Chemistry 2521, Fall Semester 2001 Sample Midterm 2 Exam Chapters 4, 5, 6 of Brown & Foote text

This exam has 5 problems on 4 pages. Make sure your copy is complete and correct. *Answer key is available in PDF format at:* www.d.umn.edu/~vzhdanki/2521/

1. (22) Using **curved arrows** and showing the structure of the **intermediate**, write **mechanisms** that account for the products in the following reactions (11 pts each):



2. (9) Arrange the following compounds in order of <u>decreasing</u> acidity:

(1)
$$CH_3CH_2CH_3$$
 (2) CH_3CH_2OH (3) $CH_3CH_2NH_2$ (4) $H - C$

3. (18) Give either the **IUPAC name** (including *E*, *Z* designation) or the **correct structure** for each of the following compounds (3 pts each).

(a)
$$\frac{7}{\frac{4}{1}} \frac{2}{\frac{2}{1}} C_{1} (E) - 1 - chloro - 4 - ethyl - 3 - fluoro - 3 - heptene$$

(b) $\frac{7}{CH_{3}CH_{2}CH_{2}} \frac{5}{CH_{3}} CH_{3} (E) - 4 - ethyl - 2, 3 - dimethyl - 3 - heptene$
 $CH_{3}CH_{2} CH_{2}CH_{2}(CH_{3})_{2} (E) - 4 - ethyl - 2, 3 - dimethyl - 3 - heptene$

(c) trans-3,4-dimethyl-2-pentene



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(d) (Z)-1,5-dibromo-2-chloro-3-ethyl-2-pentene



(e) (6E)-2,6-dimethyl-2,6-octadiene



(f) 3-ethyl-1-methylcyclopentene



4. (36; 4 pts each) Complete the following equations, showing the stereochemistry of the product when appropriate.





5. (15, 5 pts each) For each of the following questions (a)-(c) circle the item that is the correct answer.

(a) Which one of the following carbocations is the **most stable**?



(b) Which one of the following compounds is the strongest base?



(c) Which one of the following alkenes is the most stable?