Chemistry 2521, Spring Semester 2001 Sample Final Exam Chs 1-10 of Brown & Foote text

This exam has 6 problems (200 pts) on 5 pages. Make sure your copy is complete and correct. Answer key is available in PDF format at: www.d.umn.edu/~vzhdanki/2521/

1. (30) Using **curved arrows** and showing the structure of the **intermediates**, write **mechanisms** that account for the products in the following reactions (15 pts each):



2. (29) Draw three-dimensional structures of major organic products for the reaction of (R)-3methylcyclopentene with Br_2 in CCl_4 (10 pts). Assign R,S configurations for all chiral centers in each of the products. (8 pts). Using curved arrows, write a mechanism that explains stereochemistry of the reaction (11 pts).



3. (36) Complete the following equations by drawing **three-dimensional structures** with **correct stereochemistry** of the <u>major</u> organic products expected in each case (6 pts each).





4. (35) Complete the following equations by drawing structures of the **major** product(s) expected in each reaction (5 pts each).



5. (30) Give the **reagents on the arrow** that can be used to convert the reactant to the indicated product in high yield (5 pts each).



6. (40, 5 pts each) For each of the following questions (a)-(h) circle the item that is the correct answer.

(a) Which of the following compounds is the **most reactive in** $S_N 2$ reactions?

iodocyclohexane 1-iodo-2-methylhexane 1-iodo-1-methylcyclohexane 2-iodohexane 1-iodo-4-methylhexane 1-iodo-4-methylcyclohexane 5

(b) Which one of the following species is the most stable carbocation?



(c) Which of the following substituents is the best leaving group?

$$(-OT_s)$$
 $-Cl$ $-CH_3$ $-Ph$ $-OH$ $-F$ $-NH_2$

(d) Which one of the following anions is the strongest base?

 $CH_{3}NH_{2} \qquad C_{2}H_{5}O^{-} \qquad CH_{3}CH_{2}^{-} \qquad HO^{-} \qquad H_{2}O \qquad NH_{2}^{-} \qquad Br^{-} \qquad I^{-}$

(e) Which one of the following compounds has four stereoisomers?

2-bromobutane 3,4-dichlorohexane methylcyclopentane 1,1-dimethylcyclobutane 1,4-dichlorocyclohexane 2,3-dibromopentane 1,2-dibromocyclohexane

(f) Which one of the following compounds will have the highest boiling point?

$$\begin{array}{c} 0 \\ H_{3}C \\ \hline C \\ OH \end{array} \begin{array}{c} 0 \\ H_{3}O \\ \hline C \\ H \end{array} \begin{array}{c} 0 \\ CH_{3}O \\ \hline C \\ H \end{array} \begin{array}{c} CH_{3}F \\ CH_{3}O \\ \hline CH_{3}O \\$$

(g) Which one of the following compounds has the most acidic C-H bonds??

2-butyne 3-methyl-1-butene 1-buten-3-yne methane 2,3-dimethyl-2-pentene propene ethylene cyclohexene

(h) Which of the following compounds is the most stable alkene?

2-methyl-2-butene 3-methyl-1-butene 2-methyl-1-butene 2,3-dimethyl-2-pentene propene ethylene cyclohexene