1. Which of the following is the ground-state electron configuration of fluorine?

 $1s^2 2s^1$ A) $\begin{array}{ccc} B) & 1s^2 2s^2 2p_x^{-1} 2p_y^{-1} 2p_z^{-1} \\ C) & 1s^2 2s^2 2p_x^{-2} 2p_y^{-2} 2p_z^{-1} \\ D) & 2s^2 3s^2 3s^2 3p_x^{-2} 3p_y^{-1} 3p_z^{-1} \end{array}$

elements below least **2**. Which of is electronegative?



3. Which of the following correctly describes **bond polarity** in a molecule of amine?



4. Which of the following molecules or polyatomic ions is a correct Lewis structure with correct formal charges?



5. What is an approximate O-C-H bond angle in formaldehyde, $H_2C=O$, according to VSEPR?



- Quizl, Key
- 6. Which of the following molecules has dipole moment $\mu = 0 D$?



7. Which represents correct direction of dipole moment of dichloromethane?



8. Which of the following compounds has a **nonpolar** covalent bond?





$$(A) -1 (A) -1 (B) +1 (C) (C) (D) -2$$

10. Which of the following compounds has the functional group of an aldehyde?

A)
$$CH_3CO_2CH_3$$
 B) CH_3CO_2H
C) $(CH_3)_3CCHO$ D) $HOCH_2COCH_3$