## Chemistry 2541, Fall 2015 **Ouiz 4**

(30 points)

Important notes:

- Please use the provided Scantron form for your answers; you can keep the sheet with the questions and can use it as scratch paper
- Do not forget to write your name on the Scantron form
- You will not receive credit for unmarked answers or for more than one mark on answer line
- Your scores will be posted on eGradebook; graded Scantron forms will not be returned to students.

Questions 1-10: Please mark the appropriate box on the front of the Scantron form (3 pts each).

1. Which one of the following compounds is the strongest base?

A)  $C_2H_6$ 

B) NaF

C) NaBr

D) CH<sub>3</sub>NHNa

2. Which of the following is the weakest acid?

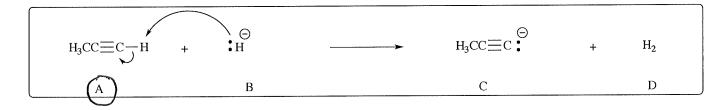
A) ICH2CO2H

B) CICH<sub>2</sub>CO<sub>2</sub>H

C) HBr

D) FCH<sub>2</sub>CO<sub>2</sub>H

3. The acid/base reaction shown in the box proceeds in the direction as shown below. Which molecule in this reaction is the **strongest** acid?



4. Which of the following represents the order of increasing acidity for compounds the box?

(1) CH<sub>3</sub>CH<sub>2</sub>I

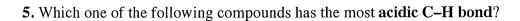
(**2**) HCl

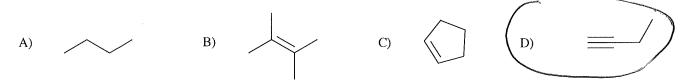
A) 1 (strongest) > 2 > 4 > 3 (weakest)

B) 2 (strongest) > 1 > 4 > 3 (weakest)

C) 2 (strongest) > 4 > 3 > 1 (weakest)

D) 4 (strongest) > 3 > 2 > 1 (weakest)



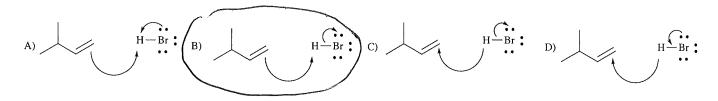


6. Which one of the following compounds is likely to react as Lewis acid?

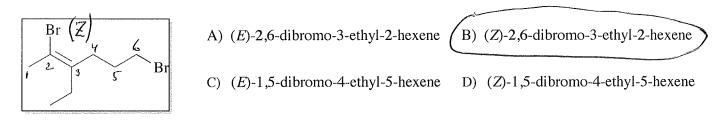


7. Which pair represents an acid and a conjugate acid for the reaction in the box?

8. Which of the following drawings correctly depicts the flow of electrons?



9. What is the IUPAC name for the compound shown in the box?



10. Which of the following compounds has index of hydrogen deficiency = 4?

