

Quiz 4

(30 points)

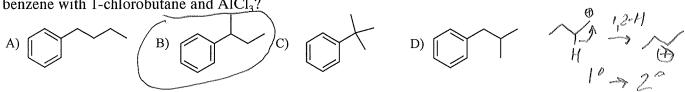


- Please use the provided Scantron form for your answers; you can keep the sheet with the questions and can use it as scratch paper
- Do not forget to write your name on the Scantron form
- You will not receive credit for unmarked answers or for more than one mark on answer line
- Your scores will be posted on eGradebook; graded Scantron forms will not be returned to students.

Questions 1-10: Please mark the appropriate box on the front of the Scantron form (3 pts each).

1. Arrange the following compounds according to decreasing order of reactivity in an electrophilic aromatic substitution reaction.

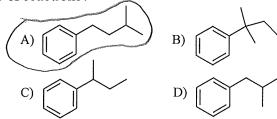
2. Which of the following compounds is a product formed by a 1,2-hydride shift in the reaction of benzene with 1-chlorobutane and AlCl₃?



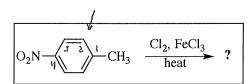
3. Which of the following is a key **intermediate** for the transformation in the box?

- **4.** Which one of the following groups will be **meta-directing** in the electrophilic aromatic substitution reaction?
- A) $-\overset{\text{f.s.}}{F}\overset{\text{f.s.}}{,}$ B) $-\overset{\text{f.s.}}{N}(CH_3)_2$ C) $-\overset{\text{f.s.}}{S}H$ D) $-CF_3$

5. What is a major **product** of the following sequence of reactions?



6. What is the name of the **product** formed in the following reaction?



- A) 2-chloro-4-nitrotoluene
- B) 3-chloro-4-methylaniline
- C) 2-chloro-4-nitrophenol
- D) p-chlorotoluene

7. What is the name of the **product** formed in the following sequence of reactions?

- A) m-bromobenzoic acid B) p-bromobenzoic acid

 - C) m-bromoethylbenzene D) p-bromoethylbenzene

8. What is the name of the major **product** of the reaction shown in the box?

$$O_2N$$
 — Cl $\frac{\text{NaOH}}{\text{water}}$?

- A) m-chlorophenol B) p-chlorophenol
- C) p-nitrophenol
- D) *m*-nitrophenol

9. Which one of the following compounds is most reactive towards nucleophilic aromatic substitution

reaction with NaOCH₃?

A)
$$O_2N$$
 O_2N
 O_2N

10. Which reagents can be used for the sequence of reactions shown in the box?

- A) (1) SO_3 , H_2SO_4 , (2) Br_2 , $FeBr_3$ B) (I) Br₂, FeBr₃, (2) SO₃, H₂SO₄ C) (1) NBS, light, (2) H₂SO₄
- D) (1) H₂SO₄, (2) NBS, light